## Note

# Synthesis of (5S,8R,9R,10S)-8,9,10-trihydroxy-2-thioxo-1,6-dioxa-3-azaspiro[4.5]decane from 1-amino-1-deoxy-D-arabino-hexulose

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The reaction of 1-amino-1-deoxy-D-arabino-hexulose with carbon disulphide at  $5^{\circ}$  yields<sup>1</sup> 5-hydroxy-5-(D-arabino-tetritol-1-yl)thiazolidine-2-thione (1), whereas 3-alkyl analogues (2), with the D-arabino, D-lyxo, D-ribo, and D-threo configurations, are prepared<sup>2,3</sup> from N-alkyl-D-glycopyranosylamines and carbon disulphide. Little attention has been directed to the syntheses of analogous spiro-nucleoside derivatives. The furanosylspiro-oxazolidine-2-thione (3) was obtained<sup>4,5</sup> by reaction of D-fructose and thiocyanic acid. The spiro structure 4 was proposed<sup>6</sup> for the product of reaction of 1-amino-1-deoxy-D-arabino-hexulose with phenyl isothiocyanate, although the compound was not isolated, and we have described<sup>7</sup> the preparation of (2R,3R,4S,5S)-3,4-dihydroxy-2-hydroxymethyl-8-methyl-6-phenyl-7- thioxo-1-oxa-6,8-diazaspiro[4,4]-nonane (5) and the 5R isomer from 1-deoxy-1-methylamino-D-lyxo-hexulose and phenyl isothiocyanate.

We now report that the reaction of 1-amino-1-deoxy-D-arabino-hexulose acetate with carbon disulphide at 75° gave (49%) (5S,8R,9R,10S)-8,9,10-trihydro-xy-2-thioxo-1,6-dioxa-3-azaspiro[4.5]decane (6),  $[a]_{\rm D} - 200^{\circ}$  (cf. +18° for 1),  $\lambda_{\rm max}$  243 nm (close to that for tetrahydro-oxazole-2-thiones<sup>8</sup> and different from that for thiazoli-

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dine-2-thiones<sup>3</sup>); the high resolution mass spectrum showed the exact atomic composition for the molecular ion. The <sup>1</sup>H-n.m.r. spectrum (see Experimental) indicated the presence of one NH and three OH groups, which confirmed the pyranoid structure. The i.r. absorptions<sup>9</sup> at 1537 ( $\delta_{\rm NH}$ ) and 1174 cm<sup>-1</sup> ( $v_{\rm C=8}$ ) and the <sup>13</sup>C-n.m.r. data confirmed the oxazole structure. The spiro carbon (C-5) resonated at 112.5 p.p.m., and was more deshielded than the similar carbon of  $2^3$  (88–90 p.p.m.), consistent with bonding to two oxygen atoms.

Conventional treatment of **6** with acetic anhydride–pyridine at room temperature or with acetic anhydride and zinc chloride at 150° yielded the tetra-acetate **7**. The <sup>1</sup>H-n.m.r. data  $(J_{9,10} 10.3; J_{7,8}, J_{7,8}, \text{ and } J_{8,9} 1.3-2.9 \text{ Hz})$  are consistent with a major  ${}^{1}C_{4}(D)$  conformation in solution in chloroform. The  $J_{7,7}$  value (13.7 Hz) accords with an axial acetoxyl group  ${}^{10}$  on C-8.

The thioacetyl structure (8) for 7 was ruled out by the i.r. (1160 cm<sup>-1</sup>) and  $^{13}$ C-n.m.r. (182.8 p.p.m.) data which are characteristic of the C=S group (cf. ~194 p.p.m. for C=O<sup>11</sup>). The  $^{1}$ H-singlet at 2.81 p.p.m. is assigned to the NAc, because its chemical shift is similar to that of NAc groups of related 2-thioxoheterocycles  $^{12.13}$  (cf. 235 p.p.m. for the acetylthio group  $^{14}$ ). This chemical shift is indicative  $^{13}$  of the conformation 9, where the methyl group is deshielded by the nearby C=S bond.

The S configuration for C-5 in 7 has been demonstrated by X-ray diffraction<sup>15</sup>.

A possible mechanism for the synthesis of 6 from fructosamine and carbon disulphide at 75° involves the formation of the thiocarbamate 10 and the isothiocyanate 11. This mechanism is related with that proposed<sup>16</sup> for the reaction of aminoalcohols and carbon disulphide.

### **EXPERIMENTAL**

General methods. — Melting points are uncorrected. Optical rotations were measured at 5461 Å, using a 1-cm cell. I.r. spectra were recorded for KBr discs. <sup>1</sup>H-N.m.r. sprectra were obtained at 200 MHz. Assignments were confirmed by decoupling experiments and H/D exchange. <sup>13</sup>C-N.m.r. spectra were recorded at 50.3 MHz for solutions in CDCl<sub>3</sub> and (CD<sub>3</sub>)<sub>2</sub>SO. Proton-decoupled APT<sup>17</sup> spectra were obtained in order to assist in signal assignments. The e.i.-mass spectrum was obtained at

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70 eV, with an ion-source temperature of 200°; accelerating voltage, 4 kV; resolution 10.000 (10% valley definition). The elemental composition of the molecular ion was determined by a peak-matching method relative to PFK.

(5S,8R, 9R,10S)-8,9,10-Trihydroxy-2-thioxo-1,6-dioxa-3-azaspiro[4.5]decane (6). — A mixture of 1-amino-1-deoxy-D-arabino-hexulose acetate<sup>18</sup> (2 g, 5.0 mmol), carbon disulphide (2 mL, 33.2 mmol), and methanol (75 mL) was stirred at 75° until the amino sugar disappeared (t.l.c.; dichloromethane-methanol, 5:1). The mixture was cooled to room temperature, filtered, and concentrated. Recrystallization of the product (1.34 g, 71%) from methanol gave 6 (0.92 g, 49%), m.p. 157–158°,  $[a]_{\rm o}^{22}$  –200° (c 1, pyridine);  $\lambda_{\rm max}^{\rm MeOH}$  243 (ε<sub>mм</sub> 18.0);  $\nu_{\rm max}$  3400 (OH), 3200 (NH), 2950, 2920, 2900 (CH), 1537 (NH), and 1174 cm<sup>-1</sup> (C = S). <sup>1</sup>H-N.m.r. data [(CD<sub>3</sub>)<sub>2</sub>SO]: δ 10.20 (bs, 1 H, NH), 5.75 (d, 1 H,  $J_{\rm H,OH}$  3.8 Hz, OH), 5.10 (d, 1 H,  $J_{\rm H,OH}$  1.8 Hz, OH), 5.00 (d, 1 H,  $J_{\rm H,OH}$  1.1 Hz, OH), 4.10–3.40 (m, 7 H, H-4,4′,7,7′,8,9,10) (the signals at δ 10.20, 5.75, 5.10, and 5.00 disappeared on addition of D<sub>2</sub>O); <sup>13</sup>C, δ 187.2 (C-2), 112.5 (C-5), 71.1 (C-8), 70.0 (2 C, C-9,10), 67.2 (C-7), and 51.9 (C-4). Mass spectrum: m/z 221.03630 (M<sup>+</sup>:; calc. for C<sub>7</sub>H<sub>11</sub>NO<sub>5</sub>S: 221.03579).

(5S,8R,9R,10S)-8,9,10-Triacetoxy-3-acetyl-1,6-dioxa-3-azaspiro [4.5]decane (7). — Conventional treatment of **6** (1.0 g, 4.5 mmol) with pyridine (5 mL) and acetic anhydride (5 mL, 53.0 mmol) gave 7 (1.12 g, 67%). After recrystallization from ethanol, the product (0.86 g, 51%) had m.p. 120–122°,  $[a]_{\rm D}^{122}-220^{\circ}$  (c 1, chloroform);  $\lambda_{\rm max}^{\rm MeOH}$  265 ( $\varepsilon_{\rm mM}$  14.0);  $\nu_{\rm max}$  2980, 2963, 2958, 2930 (CH), 1750 (CO, ester), 1700 (CO, amide), 1240, 1080 (C-O), and 1160 cm<sup>-1</sup> (C = S). <sup>1</sup>H-N.m.r. data (CDCl<sub>3</sub>):  $\delta$  5.60 (d, 1 H,  $J_{8,9}$  2.9 Hz, H-9), 5.58 (d, 1 H,  $J_{9,10}$  10.3 Hz, H-10), 5.54 (m, 1 H. H-8), 4.33 (dd, 1 H,  $J_{7,7}$  13.7,  $J_{7,8}$  1.3 Hz, H-7), 4.13 (d, 1 H,  $J_{4,4'}$  14.2 Hz, H-4), 3.98 (dd, 1 H,  $J_{7,8}$  2.1 Hz, H-7'), 3.94 (d, 1 H, H-4'), 2.81 (s, 3 H, NAc), 2.18, 2.11, and 2.01 (3 s, 9 H, 3 OAc); <sup>13</sup>C,  $\delta$  182.8 (C-2), 171.1, 167.8, 167.5, 167.3 (4 C = O), 104.4 (C-5), 69.2 (2 C, C-9,10), 68.3 (C-8), 63.6 (C-7), 53.8 (C-4), 26.0 (NCOCH<sub>3</sub>), 20.4, 20.2, and 19.7 (3 OCOCH<sub>3</sub>).

Anal. Calc. for  $C_{15}H_{19}NO_9S$ : C, 46.27; H, 4.92; N, 3.60; S, 8.24. Found: C, 46.27; H, 5.01; N, 3.71; S, 8.70.

Compound 7 (49%) was also obtained when the acetylation reaction was performed at  $150^{\circ}$  (30 min) in the presence of zinc chloride.

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